This article was downloaded by: On: *26 January 2011* Access details: *Access Details: Free Access* Publisher *Taylor & Francis* Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Liquid Crystals

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713926090

Low frequency dielectric spectroscopy on mixtures containing ferroelectric liquid crystals

S. U. Vallerien^a; F. Kremer^a; H. Kresse^b; B. Linström^b; C. Tschierske^b ^a Max-Planck-Institut für Polymerforschung, Mainz, Germany ^b Martin-Luther-Universität, Halle, Germany

To cite this Article Vallerien, S. U., Kremer, F., Kresse, H., Linström, B. and Tschierske, C.(1991) 'Low frequency dielectric spectroscopy on mixtures containing ferroelectric liquid crystals', Liquid Crystals, 9: 5, 757 – 762 **To link to this Article: DOI:** 10.1080/02678299108030388 **URL:** http://dx.doi.org/10.1080/02678299108030388

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doese should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

PRELIMINARY COMMUNICATIONS

Low frequency dielectric spectroscopy on mixtures containing ferroelectric liquid crystals

by S. U. VALLERIEN and F. KREMER* Max-Planck-Institut für Polymerforschung, Postfach 31 48, W-6500 Mainz, Germany

and H. KRESSE, B. LINSTRÖM and C. TSCHIERSKE Martin-Luther-Universität, Sektion Chemie, Postfach, O-4010 Halle, Germany

(Received 20 September 1990; accepted 13 January 1991)

Dielectric spectroscopy (frequency range 10^{-1} Hz to 10^{5} Hz, temperature range 300 K to 370 K) was employed to investigate the Goldstone mode in ferroelectric liquid-crystalline mixtures of chiral and achiral compounds. While the spontaneous polarization decreases with increasing fraction of the achiral molecules the Goldstone mode remains constant in its oscillatory strength. As expected the Goldstone mode frequency shifts to lower values with increasing amount of achiral molecules in the ferroelectric liquid-crystalline mixture.

Dielectric spectroscopy has become a versatile tool for the investigation of low molar mass [1–10] and polymeric [11] ferroelectric liquid crystals. Usually pure ferroelectric liquid crystals or a single mixture containing ferroelectric liquid crystals were studied. Whereas for display applications several ferroelectric mixtures are available and commonly used (e.g. [12]) in order to achieve a broad temperature range, a smectic A phase and a compensated cholesteric pitch for easy orientation and fast switching times can also be employed. In this preliminary communication we present a broadband dielectric spectroscopy study (frequency range: 10^{-1} Hz to 10^{5} Hz, temperature range 300 K to 370 K) of chiral/achiral systems.

The chiral system used was



C 340 K S^{*}_C 422 K S_A 423 K I.

The achiral system used was



C 328 K S_C 339 K N 362 K I.

* Author for correspondence.

0267-8292/91 \$3.00 C 1991 Taylor & Francis Ltd.

The synthesis of the chiral liquid crystals is described in [5]. The transition temperatures and phase assignments were determined by polarization microscopy and dielectric spectroscopy. The mixing was done by stirring in the isotropic phases. The dielectric measurements covered the frequency range from 10^{-1} Hz to 10^{5} Hz using a frequency response analyser (Solartron-Schlumberger FRA 1254 frequency range 10^{-4} Hz to 6×10^{4} Hz) with a high impedance preamplifier of variable gain [13–15] and a Hewlett-Packard impedance analyser (HP 4192 A frequency range from 10 Hz to 10^{7} Hz). The AC measuring field (200 V/cm) could be superimposed on an internal bias field obtained from the HP 4192 A or from the Solartron-Schlumberger FRA 1254. A nitrogen gas heating system covered the temperature range from 100 K to 450 K, the stability of the temperature adjustment was $\pm 0.02 \,\mathrm{K}$ (i.e. the resolution of the temperature measurement was 0.01 K). The temperature gradient across the sample is estimated to be less than 0.04 K. The liquid-crystalline sample was kept between two glass plates coated with polyimid and separated by a spacer ($50 \pm 4 \mu m$). The parallel rubbing direction led to the desired uniform bookshelf geometry. The alignment in the filled cell was checked by polarization microscopy. The voltage was applied by two evaporated indium-tin oxide electrodes with an area of 10 × 10 mm² (LCD-Cell from EHC, Japan).

The measured dielectric loss ε'' in its frequency and temperature dependence (see figure 1) for the 0.5 chiral (THIA)/achiral mixture reveals how much the Goldstone mode contributes in the S^{*}_C phase. Close to the phase transition from the S^{*}_C phase to the crystalline phase the Goldstone mode is shifted to lower frequency before it completely



Figure 1. Dielectric loss ε'' versus frequency and temperature for a 1:1 mixture ($x_T = 0.5$) of the chiral and achiral compound. The experimental accuracy of the dielectric measurement is ± 5 per cent.

chiral/ac	hiral liquid crystals. The spontaneous polarizat increment and the Goldstone mode frequency	ion was measured 10 K below is independent of temperature	the phase transition $I_{S_{c}S_{A}}$. The in the S_{c}^{*} phase.
Mixtures X _{chiral}	Phases	Spontaneous polarization 10 K below T_{S5SA} $P_{\text{S}}/\text{nC cm}^{-2}$	Dielectric increment $\Delta \varepsilon$ for the Goldstone mode at the temperature indicated
-	C 340 K S [*] 422 K S ^A 423 K I	125	3700±300 at 349.1 K
0-7	C 325 K S [*] 387 K S [*] 402 K I	39-5	2600 ± 210 at 350.4 K
0.5	C 318 K S [*] 371 K S [*] 380 K N [*] 386 K I	19-8	3100 ± 250 at $351\cdot0$ K
0-2	C 321 K S [*] 354 K S ^A 360 K N [*] 374 K I	5.2	2900 ± 230 at 350.4 K

Liquid-crystalline phases, spontaneous polarization and dielectric increment $\Delta \varepsilon$ of the Goldstone mode for different mixtures of l vanishes in the crystalline state, which might be a highly ordered smectic phase. Similar behaviour is found in all three mixtures (x=0.7, x=0.5, x=0.2) and in the pure thiodiazol (x=1). Diluting (see the table and figure 2) the chiral component from x=1 to x=0.2 results in a decrease of the Goldstone mode frequency from about 100 Hz for x=1 to about 2 Hz for x=0.2 whereas the dielectric strength is unaffected within the limits of the experimental accuracy. For the spontaneous polarization measured by the field reversal method decreasing values were found with decreasing content of the chiral component in the mixtures (see the table).

For the Goldstone mode frequency $v_{\rm G}$, the expression

$$\frac{1}{v_{\rm G}} = \frac{\gamma_{\rho}}{K_{\rho}q^2}$$

was derived [3 (b), 3 (c), 6 (c)] where γ_{ρ} is the rotational viscosity, K_{ρ} the coefficient in the Landau expression [1] and q the wavevector of the helical pitch Z with $q = 2\pi/Z$. Diluting the initial pure chiral substance with the achiral we expect that q should depend linearly on concentration. Thus, with a dilution factor of 5 (x=0.2), the peak frequency should be reduced by a factor of 25, or to 4 Hz, given that f(x=1)=100 Hz. This is only a factor of two higher than in the experiment, which is not striking in view of the fact that as the pitch increases the helix becomes quite susceptible to external perturbations. For example, the samples used are only 50 μ m thick, so that the sample surfaces must significantly perturb the helix structure at its larger pitches. The fact, that the frequencies for x=1 and x=0.7 are similar, is not understood.



Figure 2. Dielectric loss ε'' versus frequency for different mixtures x chiral/achiral. Due to an uncertainty in the sample thickness of ± 10 per cent the absolute accuracy in the comparison between two mixtures is only ± 10 per cent. The sample temperatures vary between 349.1 K and 351 K (see the table).

The oscillatory strength of the Goldstone mode is, in contrast, nearly unaffected within the limits of experimental accuracy, which is unexpected in view of the decreasing values of the spontaneous polarization and the different dipole moments of the chiral and achiral compounds. Obviously, even at a concentration of x=0.2, the chiral system induces a chiral S_c phase, in which the lateral dipoles of the achiral compound can contribute to the oscillatory strength of the dielectrically observable Goldstone mode. In the classical theory of the S^{*}_c phase [16–18] and in the conclusion drawn by Lalanne *et al.* from their experiments [19], free rotation of the molecules in the S^{*}_c phase.

Based on the direct observation of the high frequency rotation by broadband dielectric spectroscopy [20, 21], we could show that frequency and dielectric strength of the hindered rotation of the molecules, which has to be interpreted as a librational movement, is not influenced at the $S_C^*-S_A$ transition which is in accord with the theoretical work of Brand and Pleiner [22, 23]. For the S_A phase the angle distribution of this libration is presumably isotropic, whereas inside the S_C^* phase an anisotropic distribution also for the achiral molecules, so that their dipole moments can contribute to the oscillatory strength of the dielectric Goldstone mode. The application of a DC bias field, which normally allows the suppression of the dielectrically observed Goldstone mode inside the S_C^* phase [5, 6, 8, 11 (b)] and thereby makes the soft mode observable, is not possible in the thiodiazol compound due to the occurrence of electrochemical degradation.

The authors would like to thank Professor Dr W. Frank, Professor Dr H. R. Brand, Professor Dr H. Pleiner, Dr F. Gouda and Dr T. Geelhaar for helpful discussions. The technical assistance of G. Zak is gratefully acknowledged as well as the financial support of the Forschungsinstitut der Deutschen Bundespost.

References

- [1] MARTINOT-LARGARDE, PH., and DURAND, G., 1981, J. Phys., Paris, 42, 269.
- [2] (a) CAVA, R. J., PATEL, J. S., COLLEN, K. R., GOODBY, J. W., and RIETMANN, E. A., 1987, *Phys. Rev.* A, **35**, 4378. (b) CAVA, R. J., PATEL, J. S., and RIETMAN, E. A., 1986, *J. appl. Phys.*, **60**, 3093.
- [3] (a) LEVSTIK, A., CARLSSON, T., FILIPIČ, C., and ŽEKŠ, B., 1988, Molec. Crystals liq. Crystals, 154, 259. (b) LEVSTIK, A., CARLSSON, T., FILIPIČ, C., LEVSTIK, I., and ŽEKŠ, B., 1987, Phys. Rev. A, 35, 3527. (c) FILPIČ, C., CARLSSON, T., LEVSTIK, A., ŽEKŠ, B., BLINC, R., GOUDA, F., LAGERWALL, S. T., and SKARP, K., 1988, Phys. Rev. A, 38, 5833.
- [4] BIRADAR, A. M., WROBEL, S., and HAASE, W., 1989, Phys. Rev. A, 39, 2693.
- [5] VALLERIEN, S. U., KREMER, F., KAPITZA, H., ZENTEL, R., and FRANK, W., 1989, Phys. Lett. A, 138, 219.
- [6] (a) GOUDA, F., ANDERSSON, G., CARLSSON, T., LAGERWALL, S. T., SKARP, K., STEBLER, B., FILIPIČ, C., ŽEKŠ, B., and LEVSTIK, A., 1989, Molec. Crystals liq. Crystals Lett., 6, 151. (b) 1989, Liq. Crystals, 6, 219. (c) GOUDA, F., SKARP, K., ANDERSSON, G., KRESSE, H., and LAGERWALL, S. T., 1989, Jap. J. appl. Phys., 28, 1887.
- [7] OZAKI, M., NAKOA, K., HATAI, T., and YOSHINO, K., 1989, Liq. Crystals, 5, 1219.
- [8] GLOGOROVA, M., and PAVEL, J., 1989, Liq. Crystals, 6, 325.
- [9] BENGUIGUI, L., 1982, J. Phys., Paris, 43, 915.
- [10] LEGRAND, G., PARNEIX, J. P., M'KADMI, A., FINK, N. H., DESTRADE, C., SALLENEUVE, C., and ISAERT, N., 1988, Ferroelectrics, 84, 249.
- [11] (a) VALLERIEN, S. U., ZENTEL, R., KREMER, F., KAPITZA, H., and FISCHER, E. W., 1989, Makromolek. Chem. rap. Commun., 10, 333. (b) Proceedings of the 2nd FLC-Conference, Göteborg, Sweden, Ferroelectrics (in the press). (c) VALLERIEN, S. U., KREMER, F., KAPITZA, H., ZENTEL, R., and FISCHER, E. W., 1989, Proceedings of the 7th International Meeting on Ferrelectricity (IMF-7), Saarbrücken FRG, Ferroelectrics, 109, 273.

- [12] MERCK: 1989, Ferroelectric Smectic Mixtures Catalogue, Darmstadt.
- [13] PUGH, J., and RYAN, J. T., 1979, *IEE-Conference on Dielectric Materials, Measurements and Applications*, Aston (England), IEE Conference Publication, No. 177, p. 404.
- [14] (a) VALLERIEN, S. U., KREMER, F., HUESER, B., and SPIESS, H. W., 1989, Colloid Polymer Sci.,
 267, 583. (b) VALLERIEN, S. U., KREMER, F., and BOEFFEL, C., 1989, Liq. Crystals, 4, 79.
- [15] KREMER, F., BOESE, D., MEIER, G., and FISCHER, E. W., 1989, Prog. Colloid Polymer Sci., 80, 129.
- [16] MEYER, R. B., 1977, Molec. Crystals liq. Crystals, 40, 33.
- [17] BERESNEV, C. A., and BLINOV, L. M., 1981, Ferroelectrics, 33, 129.
- [18] LAGERWALL, S. T., and DAHL, I., 1984, Molec. Crystals liq. Crystals, 114, 151.
- [19] LALANNE, J. R., BUCHERT, J., DESTRADE, C., NGUYEN, H. T., and MARCCRON, J. P., 1989, Phys. Rev. Lett., 62, 3046.
- [20] VALLERIEN, S. U., KREMER, F., GEELHAAR, T., and WÄCHTLER, A. E., 1990, Phys. Rev. A, 42, 2482.
- [21] KREMER, F., VALLERIEN, S. U., KAPITZA, H., ZENTEL, R., and FISCHER, E. W., 1990, Phys. Rev. A, 42, 3667.
- [22] BRAND, H. R., and PLEINER, H., 1990, Phys. Rev. Lett., 64, 1309.
- [23] BRAND, H. R., and PLEINER, H., 1991, Phys. Rev. A (in the press).